



A Practical Guide for XMVB Calculations

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Outline

- General introduction
- How to write input files?
- How to get information from output files?

General introduction



An Ab Initio Non-orthogonal Valence Bond Program

Version 2.1

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Distributions of XMVB 2.1

Module distribution

→ A module embedded in GAMESS-US

Stand-alone distribution

→ A stand-alone program

How to Run An XMVB Job?

For Stand-alone distribution

(1) **preint ***.inp** ← Get integrals for VB calculation

(2) **XMVB ***.xmi** ← VB calculation

***.xmo

***.orb

***.den

Useful documents

XMVB manual

<http://www.xmvb.org/art.php?id=8>

Online Tutorials

<http://www.xmvb.org/art.php?id=4>

How to write input files?

How to write INT File

```
RHF cc-pvdz
```

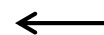
```
0 1
```

```
H 0.000000 0.000000 0.000000
```

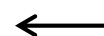
```
H 0.000000 0.000000 0.740000
```



HF Method; basis set

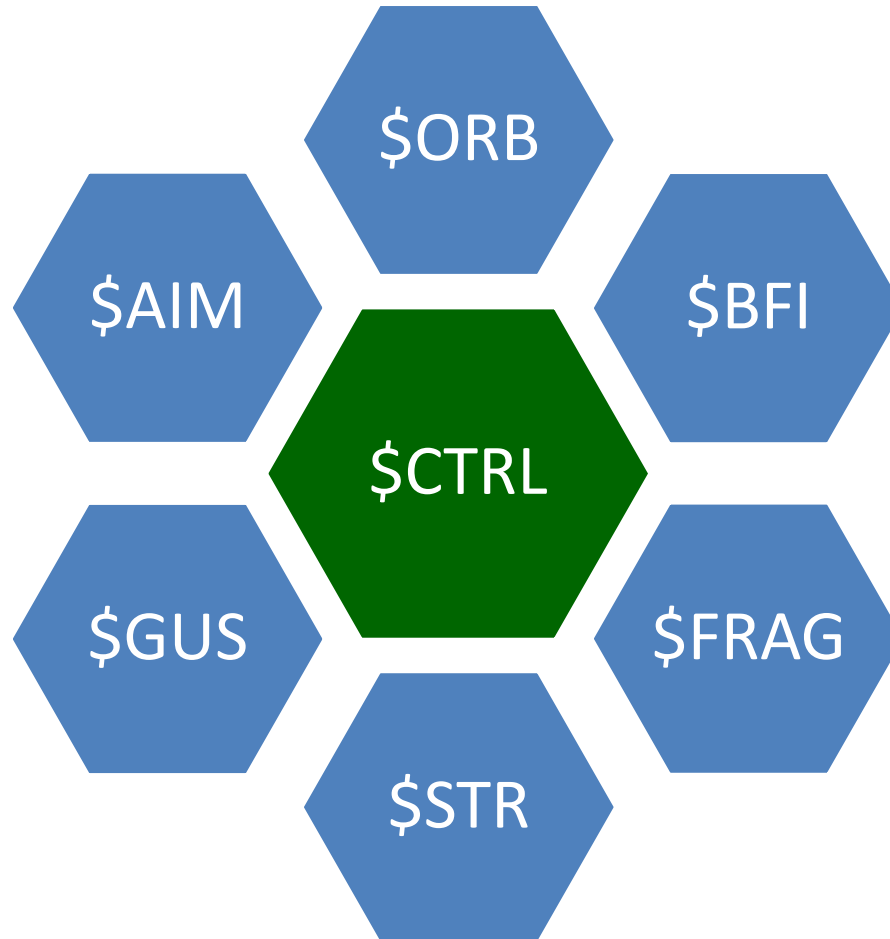


Charge; multiplicity



Molecular coordinates

Structure of XMI File

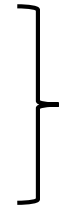
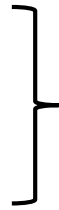


Optional

Necessary

Example 1. H₂ molecule

```
$ctrl  
nstr=3 iscf=5 nao=2 nae=2  
orbtyp=hao frgtyp=sao  
$end  
$str  
1 2  
1 1  
2 2  
$end  
$frag  
1 1  
spz 1  
spz 2  
$end  
$orb  
1 1  
1  
2  
$end
```



```

$ctrl
nstr=3 iscf=5 nao=2 nae=2
orbtyp=hao frgtyp=sao

```

```

$end
$str
1 2
1 1
2 2
$end
$frag
1 1
spz 1
spz 2
$end
$orb
1 1
1
2
$end

```



iscf : orbital optimization algorithm
 (VBSCF, BOVB)
 When iscf=5, the two keywords are
 necessary
nao: number of active orbitals
nae: number of active electrons

iscf	Gradient	Performance	Required keywords
1	Numerical	Stable, slow	/
2	Analytical	Fast	
3	Numerical	Stable, slow	
4	Numerical	Stable, very slow	
5	Analytical	Very Fast	nao, nae
6	Full Hessian	Slow	nao, nae

```

$ctrl
nstr=3 iscf=5 nao=2 nae=2
orbtyp=hao frgtyp=sao

```

```
$end
```

```
$str
```

```
1 2
```

```
1 1
```

```
2 2
```

```
$end
```

```
$frag
```

```
1 1
```

```
spz 1
```

```
spz 2
```

```
$end
```

```
$orb
```

```
1 1
```

```
1
```

```
2
```

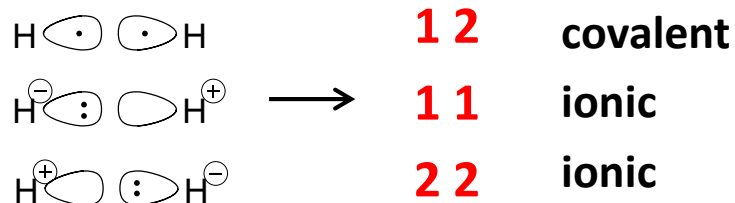
```
$end
```

nstr =3, the number of structures
 \$str : describing VB structure or
 determinant.

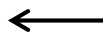
In \$str, a typical structure is written as:

```
a a b b c c ... m n
```

a b c : double occupied orbitals (inactive)
 m n : single occupied orbitals (active)



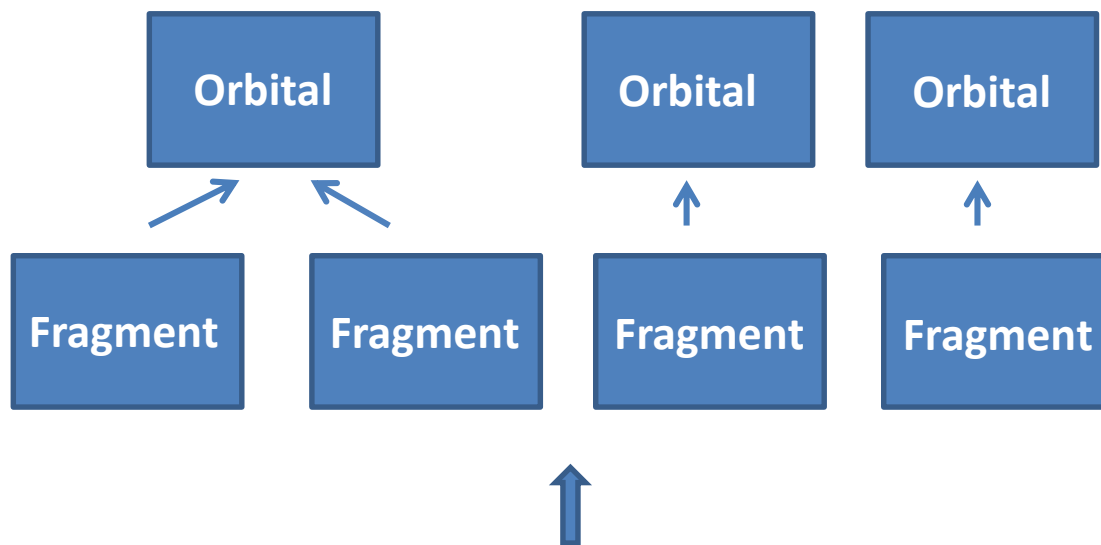
```
$ctrl  
nstr=3 iscf=5 nao=2 nae=2  
orbtyp=hao frgtyp=sao
```



```
orbtyp=hao,  
Hybrid atomic orbitals.  
localized non-orthogonal orbitals
```

```
$end  
$str  
1 2  
1 1  
2 2  
$end  
$frag  
1 1  
spz 1  
spz 2  
$end  
$orb  
1 1  
1  
2  
$end
```

How to define orbitals?



Primitive basis functions

Fragments are a series of basis function sets.

← **frgtyp=sao**, fragments are defined with symmetrized **a**tomical **o**rbital.

```
$ctrl  
nstr=3 iscf=5 nao=2 nae=2  
orbtyp=hao frgtyp=sao
```

```
$end
```

```
$str
```

```
1 2
```

```
1 1
```

```
2 2
```

```
$end
```

```
$frag
```

```
1 1
```

```
spz 1
```

```
spz 2
```

```
$end
```

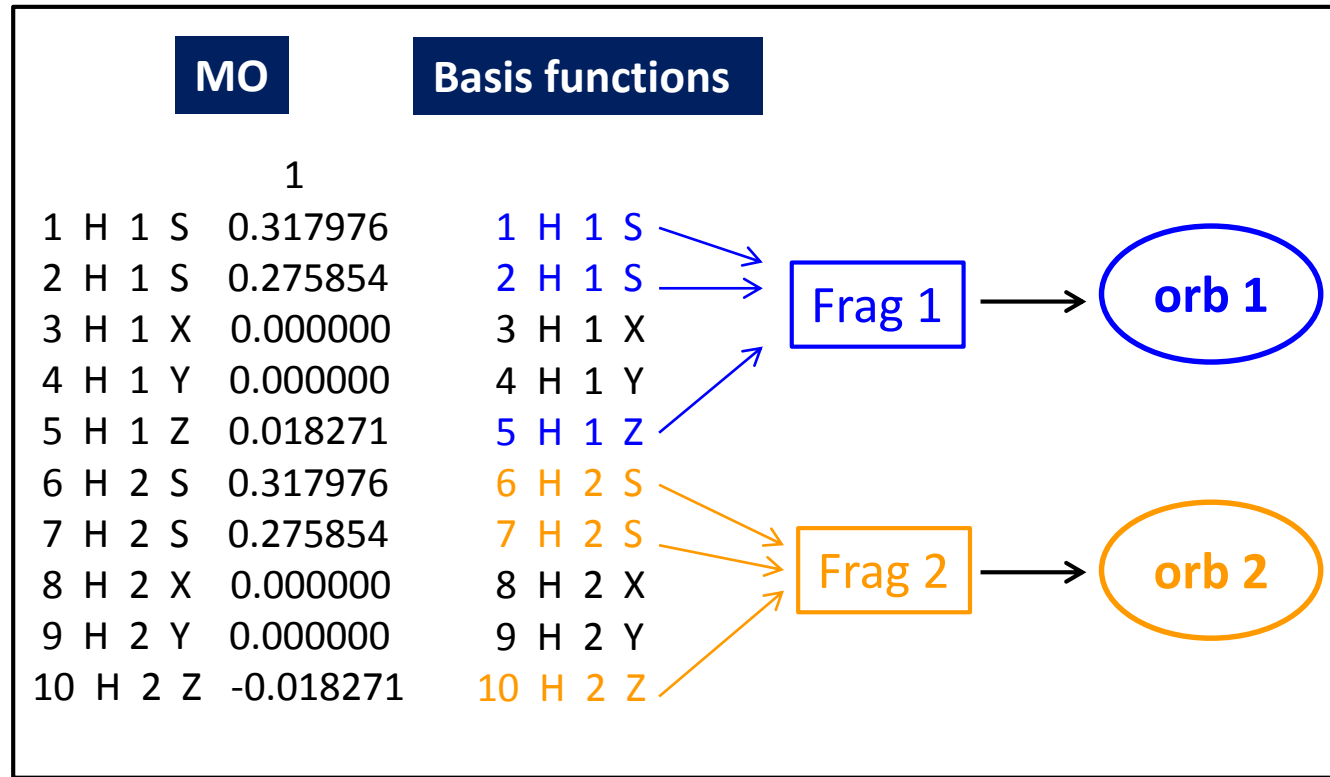
```
$orb
```

```
1 1
```

```
1
```

```
2
```

```
$end
```




```
$ctrl
nstr=3 iscf=5 nao=2 nae=2
orbtyp=hao frgtyp=sao
```

```
$end
```

```
$str
```

```
1 2
```

```
1 1
```

```
2 2
```

```
$end
```

```
$frag
```

```
1 1
```

```
spz 1
```

```
spz 2
```

```
$end
```

```
$orb
```

```
1 1
```

```
1
```

```
2
```

```
$end
```

\$frag: description for fragments

Line 1: the number of atoms in each fragment.

Line 2 : the basis functions of s and pz in atom 1

Line 3 : the basis functions of s and pz in atom 2.

\$orb: description for VB orbitals

Line 1: the number of fragments in each orbital.

Line 2: orbital 1 is described by fragment 1

Line 3: orbital 2 is described by fragment 2.

Summary (1)

1. The assignment of orbital optimization algorithm (iscf)
2. The writing of VB structures with \$str
3. The determination of VB orbitals with 'Frgtyp=sao'

Example 2. HF molecule

hf.int

RHF basis set

0 1

H 0.000000 0.000000 0.000000

F 0.000000 0.000000 0.917000

HF molecule, 3 structures

hf.xmi

```
$ctrl  
str=full nae=2 nao=2 iprint=3 orbtyp=hao  
frgtyp=atom iscf=5 guess=mo  
$end  
$orb  
1 1 1 1 1 1  
2  
2  
2  
2  
2  
1  
$end  
$gus  
1 1  
2 2  
3 4  
4 5  
5 3  
6 3  
$end
```



HF molecule, 3 structures

hf.xmi

\$ctrl

str=full nae=2 nao=2 iprint=3 orbtyp=hao
frgtyp=atom iscf=5 guess=mo

\$end

\$orb

1 1 1 1 1 1

2

2

2

2

2

1

\$end

\$gus

1 1

2 2

3 4

4 5

5 3

6 3

\$end

Automatically generate structures based on the active orbitals and electrons.

str = full => generate all VB structures

cov => only covalence

lon => only ionic structures

nao = m, the final m orbitals are selected as active orbitals.

nae = n, the number of active VB electrons

iprint=3, full print-out message.

Using str, \$str can be absent.

HF molecule, 3 structures

hf.xmi

\$ctrl

```
str=full nae=2 nao=2 iprint=3 orbtyp=hao
frgtyp=atom iscf=5 guess=mo
```

\$end

\$orb

1 1 1 1 1 1

2

2

2

2

2

1

\$end

\$gus

1 1

2 2

3 4

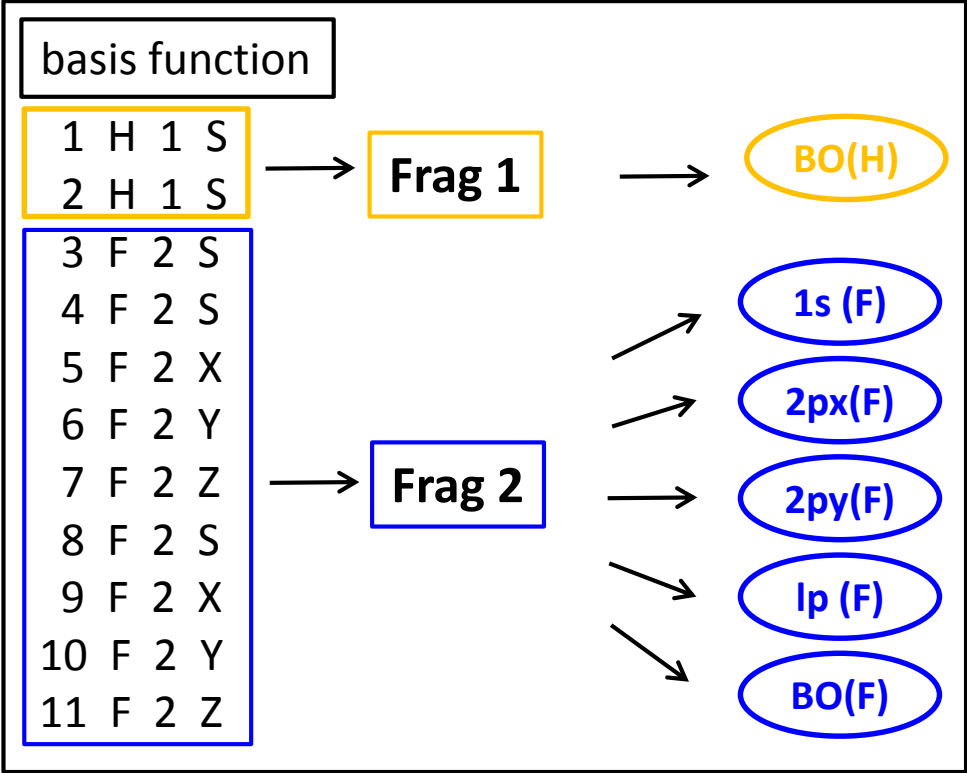
4 5

5 3

6 3

\$end

frgtyp=atom
The fragments of system are defined with atoms.
Fragtyp=atom and without \$frag means that the orbitals will be determined by the atom directly.

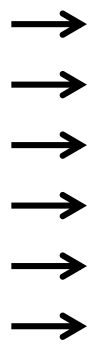


HF molecule, 3 structures

hf.xmi

\$ctrl
str=full nae=2 nao=2 iprint=3 orbtyp=hao
frgtyp=atom iscf=5 guess=mo
\$end

\$orb
1 1 1 1 1 1
2
2
2
2
2
1
\$end

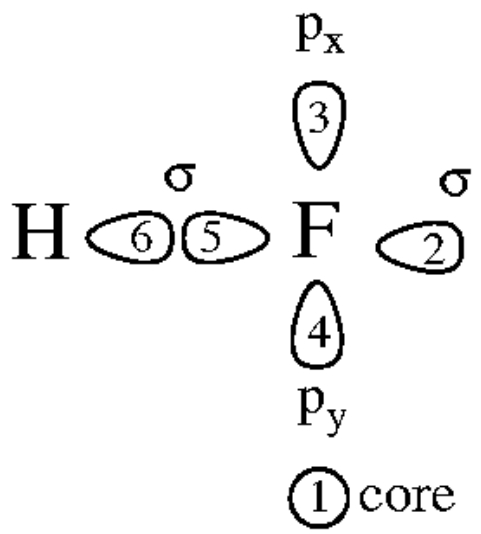


VB orb	
1 =>	frag 2
2 =>	frag 2
3 =>	frag 2
4 =>	frag 2
5 =>	frag 2
6 =>	frag 1



\$orb is required when orbtyp=hao
Line 1 shows that there are six orbitals.
While the first 5 orbitals are described by the basis functions of F atom. (frag 2)
The last orbitals are described by the basis functions of H atom. (frag 1)

\$gus
1 1
2 2
3 4
4 5
5 3
6 3
\$end



HF molecule, 3 structures

hf.xml

\$ctrl

```
str=full nae=2 nao=2 iprint=3 orbtyp=hao
frgtyp=atom iscf=5 guess=mo
```

\$end

\$orb

1 1 1 1 1 1

2

2

2

2

2

1

\$end

\$gus

1 1

2 2

3 4

4 5

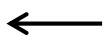
5 3

6 3

\$end

guess=mo : Initial guess of VB orbitals will be obtained directly from molecular orbitals

VB orb		MO
1	1s orbital of F atom	1
2	Lone pair of F atom	2
3	px pair of F atom	4
4	py pair of F atom	5
5	H-F bonding orbital on F atom	3
6	H-F bonding orbital on H atom	3



				1	2	3	4	5
				-26.2757	-1.5897	-0.7391	-0.631	-0.631
				A	A	A	A	A
1	H	1	S	0.000301	0.132423	-0.293056	0.000000	0.000000
2	H	1	S	0.001771	-0.013969	-0.107756	0.000000	0.000000
3	F	2	S	0.995513	-0.230516	-0.061755	0.000000	0.000000
4	F	2	S	0.021368	0.497220	0.137028	0.000000	0.000000
5	F	2	X	0.000000	0.000000	0.000000	0.654168	0.091145
6	F	2	Y	0.000000	0.000000	0.000000	-0.091145	0.654168
7	F	2	Z	-0.001488	-0.084307	0.548328	0.000000	0.000000
8	F	2	S	-0.005826	0.529428	0.244193	0.000000	0.000000
9	F	2	X	0.000000	0.000000	0.000000	0.485512	0.067646
10	F	2	Y	0.000000	0.000000	0.000000	-0.067646	0.485512
11	F	2	Z	0.001400	-0.047087	0.349640	0.000000	0.000000

Summary (2)

1. Automatically generate structures (str)
2. The determination of VB orbitals with 'Frgtyp=atom'
3. The usage of 'guess=mo' with \$gus.

Example 3. L-BOVB calculation for HF

Using VBSCF orbitals as guess orbitals

```
cp hf.orb hf-bovb.gus
```

Preparing GAMESS and XMVB input files for BOVB

```
hf-bovb.xmi
```

Running XMVB

hf-bovb.xmi

HF molecule, 3 structures

\$ctrl

str=full nae=2 nao=2 **bovb**

iprint=3 orbtyp=hao frgtyp=atom **iscf=2**

guess=read

\$end

\$orb

1 1 1 1 1 1

2

2

2

2

2

1

\$end



iscf = 4, 5,6 can not be used for BOVB calculations.

iscf=2 is recommended for BOVB

Example 4, benzene molecule

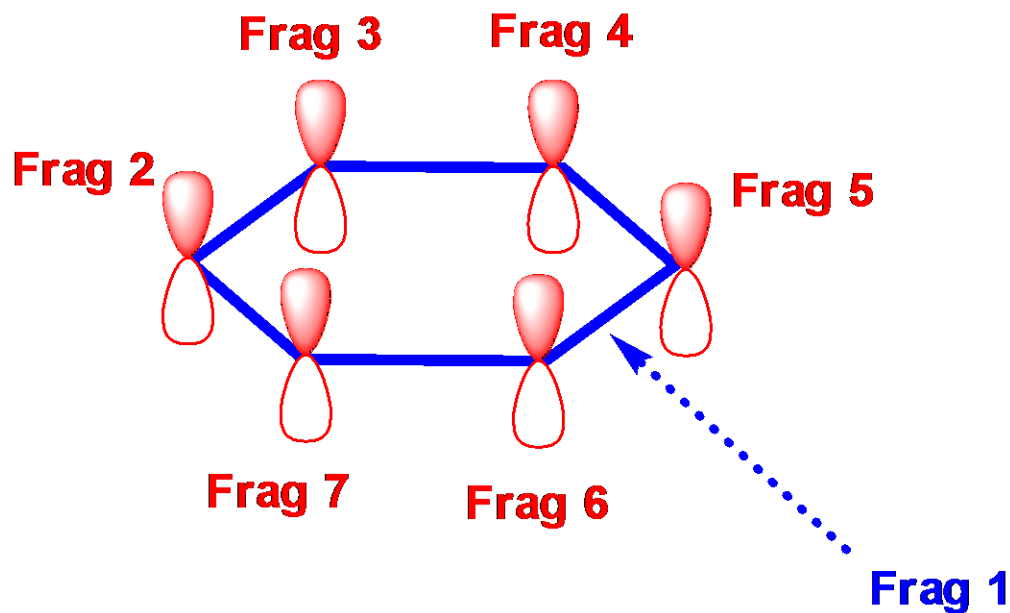
ben.int

RHF 6-31G*

0 1

C	0.6993466831	1.2113039873	0.0000000000
C	-0.6993466831	1.2113039873	0.0000000000
C	-1.3986933663	0.0000000000	0.0000000000
C	-0.6993466831	-1.2113039873	0.0000000000
C	0.6993466831	-1.2113039873	0.0000000000
C	1.3986933663	0.0000000000	0.0000000000
H	1.2427030147	2.1524247602	0.0000000000
H	-1.2427030147	2.1524247602	0.0000000000
H	-2.4854060294	0.0000000000	0.0000000000
H	-1.2427030147	-2.1524247602	0.0000000000
H	1.2427030147	-2.1524247602	0.0000000000
H	2.4854060294	0.0000000000	0.0000000000

For C₆H₆, If the π bonding is focused, the orbitals for the bonding in XY plane (σ bonding), can be considered as one fragment. The π bonding can be considered as six fragments.



ben.xmi (part1)

benzene

```
$ctrl  
str=full nao=6 nae=6  
iscf=5 iprint=3 guess=auto  
orbtyp=hao frgtyp=sao
```

\$end

```
$frag  
12 1*6  
spxpydxxdydzzdxy 1-12  
pzdxxdyz 1  
pzdxxdyz 2  
pzdxxdyz 3  
pzdxxdyz 4  
pzdxxdyz 5  
pzdxxdyz 6  
$end
```



nao =6, the final 6 orbitals are selected as active orbitals.

nae=6, the number of active VB electrons

str = full => generate all VB structures

cov => only covalence

lon => only ionic structures

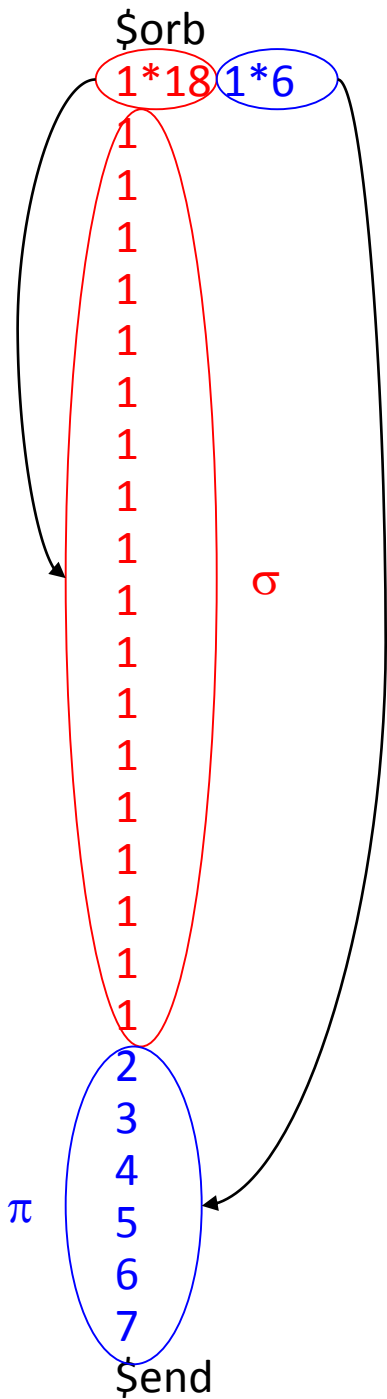


frgtyp=sao, fragments are defined with symmetrized atomic orbitals.

Here “s” means basis function S, “pxy” means basis function Px and Py. “dxxdyy” means Dxx, Dyy,.

1-12 means the appointed basis sets of atoms 1~12

ben.xmi (part2)



There are 24 VB orbitals, the first 18 orbitals are described by frag 1; the remain 6 orbitals are described by frag 2 ~ frag 7 respectively.

**How to get information
from output files?**

Information of XMO file

Input message

→ Input file, Integrals, Guess etc

SCF iterative procedure (final energy)

→ Iteration ** E = **.** DE = **
VBSCF converged in ** iterations
Total Energy: **

S_{ij} and H_{ij} Matrices

Information of VB wavefunction

→ Coefficients of structures and determinants
Weights of structures
Optimized orbitals

Properties and analysis

→ Bond order, atomic charge, dipole moments

Optional

Default

The VB calculation message in gamess output file is not correct

Total Energy: -100.03252880
First Excited: -98.677762
The Last Change in Energy: -0.000000
Number of Iteration: 32

SCF procedures (final energy)

***** MATRIX OF OVERLAP *****

	1	2	3
1	1.000000	-0.694887	-0.694887
2	-0.694887	1.000000	0.318276
3	-0.694887	0.318276	1.000000

S_{ij} and H_{ij} matrices

If iscf =5, the S_{ij} and H_{ij} matrices only contain the information of active electrons. If you want to obtain the total H_{ij} of all electrons, please use iscf = 1, 2 and 3

***** MATRIX OF HAMILTONIAN *****

	1	2	3
1	-2.554856	2.041815	1.827629
2	2.041815	-2.247238	-1.090411
3	1.827629	-1.090411	-1.685275

***** WEIGHTS OF STRUCTURES *****

1	0.64395	*****	1:4	5	6
2	0.30554	*****	1:4	5	5
3	0.05051	*****	1:4	6	6

Weights of VB structures

	1	2	3	4	5	6
1	0.000000	0.000000	0.000000	0.000000	0.000000	-0.835424
2	0.000000	0.000000	0.000000	0.000000	0.000000	-0.227552
3	1.012748	-0.084340	0.084340	-0.084340	0.049531	0.000000
4	-0.025976	0.300588	-0.300588	0.300588	-0.165593	0.000000
5	0.000000	-0.020440	-0.477373	-0.456933	0.000000	0.000000
6	0.000000	0.539422	0.252010	-0.287413	0.000000	0.000000
7	-0.010873	0.069765	-0.069765	0.069765	0.701583	0.000000
8	-0.054032	0.316476	-0.316475	0.316476	-0.042957	0.000000
9	0.000000	-0.015134	-0.353442	-0.338309	0.000000	0.000000
10	0.000000	0.399383	0.186585	-0.212797	0.000000	0.000000
11	-0.006949	0.056493	-0.056493	0.056493	0.419452	0.000000

Optimized orbitals

***** POPULATION AND CHARGE *****

ATOM	MULL.POP.	CHARGE	LOW.POP.	CHARGE
1 H	0.744964	0.255036	0.855733	0.144267
2 F	9.255036	-0.255036	9.144267	-0.144267

XMVB atomic
population analysis

***** BOND ORDER *****

ATOM 1	ATOM 2	DIST	BOND ORDER
1 H	2 F	0.900	0.899

The logo features a stylized blue icon on the left, resembling a bird or a wing, followed by the text 'XMVB - GUI' in a bold, blue, sans-serif font. The background is a dark blue gradient.

XMVB - GUI

A Graphical User Interface for XMVB Program